

Phone: +442081445350

www.chemistryonlinetuition.com

Email:asherrana@chemistryonlinetuition.com

CHEMISTRY INORGANIC CHEMISTRY II

Level & Board	AQA (A-LEVEL)
TOPIC:	AMINO ACIDS, PROTEIN AND DNA
PAPER TYPE:	QUESTION PAPER - 1
TOTAL QUESTIONS	10
TOTAL MARKS	31

ChemistryOnlineTuition Ltd reserves the right to take legal action against any individual/ company/organization involved in copyright abuse.

Amino Acids, Protein and DNA

- 1. Consider the amino acid serine.
 - Н | H₂N—С—СООН | СН₂ОН
 - (a)Draw the structure of the product formed when serine reacts with an excess of CH₃Br

(1)

(b)Draw the structure of the dipeptide formed by two molecules of serine.

(1)

2. Which is the structure of a zwitterion of an amino acid?

 $H_3N^{+} - CH - COO^{-}$ $H_2C - CH_2 - CH_2 - CH_2 - ^{+}NH_3$

am Sorry !!!!!

Α.

$$H_{3}N^{*} - CH - COO^{-}$$

$$H_{2}C - COO^{-}$$

$$H_{2}N - CH - COO^{-}$$

$$H_{2}C - ^{*}OH_{2}$$
C.
$$H_{3}N^{*} - CH - COO^{-}$$

$$H_{2}C - ^{*}OH_{2}$$
D.

3. Leucine (2-amino-4-methylpentanoic acid) is a naturally occurring α -amino acid that is often used in protein supplements.

Leucine has a structural formula of (CH₃)₂CHCH₂CH(NH₂)COOH.

(a) State the general formula of an α -amino acid.

(1)

(1)

(b)Draw a displayed formula of leucine.



4. When the dipeptide shown below is heated under acidic conditions, a single amino acid is produced.

I am Sorry !!!!!



(a)Name this amino acid.

- (1)
- (b)Draw the structure of the amino acid species present in the acidic solution.

(1)

5. Which structure shows part of a peptide link in a protein?



6. Kevlar is a very tough polymer made from 1,4-diaminobenzene and benzene-1,4-dicarboxylic acid.

(a) State a use for Kevlar.

(b)Describe the polymerisation reaction that forms Kevlar. Include in your answer:

- an explanation of the type of polymerisation involved
- an equation for the reaction
- a repeat unit to show the structure of Kevlar.

7. The amino acid R is shown below.



(a) Draw the structure of the zwitterion formed by R.

(1)

(5)

(b)Draw the structure of the major organic product formed when an excess of R is reacted with bromomethane.

(c)Name the mechanism of the reaction which results in the formation of the product.

(1)

8. The following figure shows a simplified representation of the arrangement of some amino acids in a portion of a protein structure in the form of an α -helix.



(a)Name the type of protein structure in the figure.



(b)Explain the origin of the interaction represented by the dotted lines in the figure above.

I am Sorry !!!!!

9. Which structure is formed by aspartic acid in solution at pH 12?



(1)

10. Leucine (2-amino-4-methylpentanoic acid) is a naturally occurring α -amino acid that is often used in protein supplements.

Leucine has a structural formula of $(CH_3)_2CHCH_2CH(NH_2)COOH$. Leucine can be obtained from a source of protein such as meat.

(a)State suitable reagents and conditions to break down a protein into amino acids.

I am Sorry !!!!!

(2)

(b)State the type of reaction occurring.

(1)

www.chemistryonlinetuition.com

🖂 asherrana@chemistryonlinetuition.com



- Founder & CEO of Chemistry Online Tuition Ltd.
- Tutoring students in UK and worldwide since 2008
- CIE & EDEXCEL Examiner since 2015
- · Chemistry, Physics, and Math's Tutor

CONTACT INFORMATION FOR CHEMISTRY ONLINE TUITION

- · UK Contact: 02081445350
- International Phone/WhatsApp: 00442081445350
- Website: www.chemistryonlinetuition.com
- Email: asherrana@chemistryonlinetuition.com
- Address: 210-Old Brompton Road, London SW5 OBS, UK