



CHEMISTRY ONLINE
— **TUITION** —

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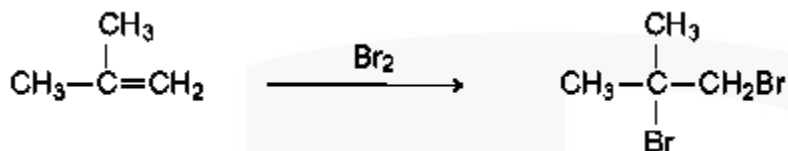
Level & Board	AQA (A-LEVEL)
TOPIC:	ALKENES
PAPER TYPE:	SOLUTION - 2
TOTAL QUESTIONS	10
TOTAL MARKS	32

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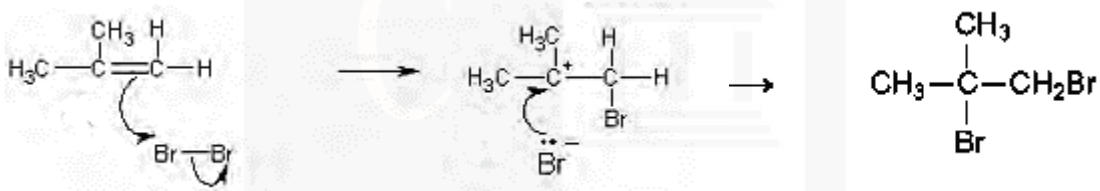
Alkenes - 3

1.

Name of Mechanism: Electrophilic addition



Mechanism:



(5)

2. B

(1)

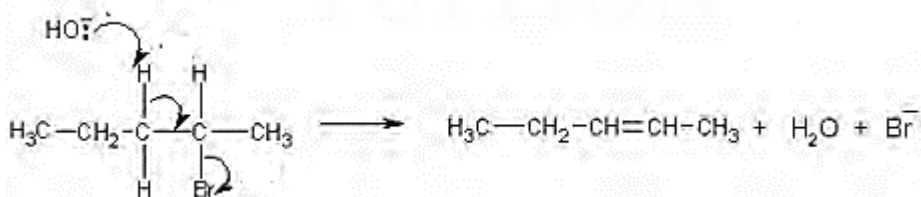
3.

(a)

Name of Mechanism: Elimination



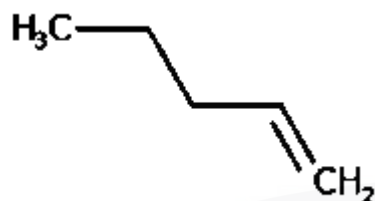
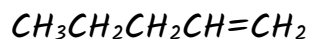
Mechanism:



(4)

(b)

Structure of isomeric alkene i.e pent-1-ene



(1)

4. C

(1)

5.

(a)

For the process:

Reagent: Hydrogen gas (H_2)

Conditions: Nickel (Ni) catalyst

Temperature: $100-200^\circ\text{C}$ or heat

(2)

(b)

Difference in Structure:

Soft Margarine:

Soft margarine is less hydrogenated, containing more $\text{C}=\text{C}$ bonds, and is more unsaturated than hard margarine.

Difference in Melting Point:

Soft Margarine:

Soft margarine has a lower melting point compared to hard margarine.

(4)

6. A

(1)

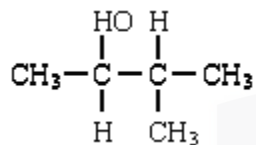
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7.

(a)

Name of alcohol:

3-methylbutan-2-ol



(1)

(b)

Elimination/dehydration reaction is involved in the formation of the two alkenes.

(1)

(c)

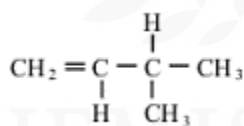
Reagent X can be:

 H_2SO_4 or H_3PO_4

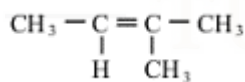
(2)

(d)

Structural formulae of the two isomeric alkenes can be represented as:



Alkene 1:



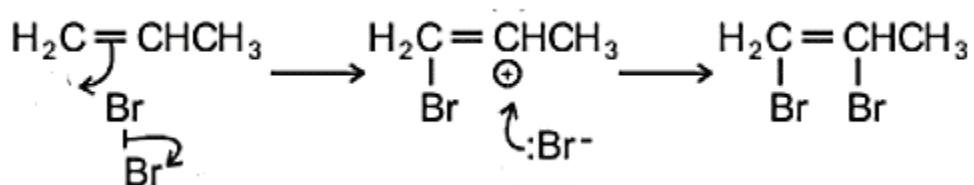
Alkene 2:

(2)

8. D

(1)

9.

Mechanism:*Name of product: 1,2-dibromopropane*

(5)

10. A

(1)



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DR. ASHAR RANA
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- Founder & CEO of Chemistry Online Tuition Ltd.
- Completed Medicine (M.B.B.S) in 2007
- Tutoring students in UK and worldwide since 2008
- CIE & EDEXCEL Examiner since 2015
- Chemistry, Physics, Math's and Biology Tutor

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