Carbohydrates and Lipids

Mark Scheme 1

Level	International A Level
Subject	Biology
Exam Board	CIE
Topic	Biological Molecules
Sub Topic	Carbohydrates and Lipids
Booklet	Theory
Paper Type	Mark Scheme 1

Time Allowed: 42 minutes

Score : /35

Percentage : /100

Grade Boundaries:

A*	Α	В	С	D	E	U
>85%	′77.5%	70%	62.5%	57.5%	45%	<45%

1 (a)

structural feature	triglyceride	phospholipid			
phosphate (group)/contains phosphorus	×	✓			
nitrogen	×	✓			
charged/polar	×	✓			
(number of) fatty acids	3	2			
number of ester bonds	3	2			
number of phosphate ester bonds	0				
award one mark for any of the following comparisons					
number of double bonds (in hydrocarbon chain)	0	1	These are		
number of saturated fatty acids/ORA	3	1	alternatives and		
presence of double bonds	×	✓			
presence of unsaturated fatty acids	×) 		

[max 2]

- (b) answer may be phrased in the context of amylase/trypsin ignore anything before Golgi, e.g. shuttle vesicles from RER
 - 1 vesicles, form from/'pinch off', Golgi (apparatus/body/complex);
 - 2 vesicles moves, through cytoplasm/to cell (surface) or plasma membrane;
 - 3 role of cytoskeleton/microtubules in movement of vesicles;
 - 4 energy/ATP, is required (movement of vesicles/fusion with membrane);
 - vesicle fuses with/AW, cell (surface)/plasma, membrane;I bind/attachA join/merge/becomes part of
 - 6 exocytosis/vesicle 'opens up' so that enzyme molecules are released;
 - 7 ref to fluid nature of, membranes/phospholipid bilayer, that makes this possible;

[max 4]

(c)

role of water	property of water	
solvent for glucose and ions	dipolar/polar; A description of polarity of water	
transport in the xylem	hydrogen bonding; I cohesion/adhesion	
helps to decrease body temperature in humans	high latent heat of vapourisation/ high specific heat (capacity)/ high enthalpy heat of vapourisation/ lots of energy required for evaporation;	

[3]

[Total: 9]



2 (a (i) -H and -OH indicated; A -OH on end of amylose and -H on alpha glucose water eliminated/condensation; A dehydration oxygen bridge/glycosidic bond, drawn between C1 and C4;

If the whole glucose molecule and/or the end of the amylose molecule has not been drawn, then only award mp3 if C1 and C4 are indicated in some way, e.g. by numbering them or putting in the hydrogens. [3]

(ii) (1,4/1,6) glycosidic; A glucosidic A phonetic spelling of glycosidic

— TUITION —

[1]

(b)

feature	а	glycogen	cellulose
type of glucose	α -glucose	α -glucose	β-glucose;
branched or unbranched molecule	unbranched		unbranched/not branched;
role in organisms	energy storage	energy storage	structural/(component of) cell walls/tensile strength/dietary fibre/roughage; I support

[3]

(c) (i) maltase and maltose must be correctly referenced ignore references to reversible/irreversible

(ascorbase) binds to/fits into/enters active site; complementary (shape) to active site; so substrate/maltose, cannot enter/cannot bind;

A no/few, ES complex

A prevents formation of ES complexes

A ascorbase forms enzyme inhibitor complex

competes with substrate/competitive inhibition; slows the (rate of), digestion/hydrolysis/breakdown, of maltose;

R 'stops the reaction'

R if in context of starch

alternative answer if candidates assume ascorbase is an enzyme: ascorbase, breaks down/digests/hydrolyses, maltase;

A ascorbase destroys the active site of maltase so no enzymes to digests maltose; slows/stops, reaction/digestion/hydrolysis/breakdown, of maltose;

[max 3]

(ii) inhibits/slows down/prevents, breakdown/(catalysing) hydrolysis/digestion, of maltose (to glucose); I starch

less glucose is absorbed/passes across membranes/enters blood;

[Total: 12]

[2]

3	(a	gly	cosic	dic; A glucosidic	[1]
	(b)	B =	mal	nalose; Itose; obiose; rose;	[max 3]
(c)		ide	a of	separation / barrier / AW, from surroundings / external environment ;	
	2	reg	ulate	es / controls / AW, entry / exit, substances / named substances ;	
	3			recognition of self (antigens) / cell recognition / avoids cell destruction s / AW;	/ act as
	4	allo	ws t	oinding of / receptors for, hormones / signal molecules / neurotransmitters / ar	ntigens;
	5	cell	to c	ell adhesion ;	
	6	loca	ation	for enzymes / multi-enzyme systems / enzyme pathways;	
	7			e.g. idea of flexibility (for some cells, coproteins / glycolipids, form H bonds with water for stability	[max 3]
	(d)	(i)	1	active site has, specific / particular, shape ;	
	(α)	(')	2		
				complementary to substrate; A substrate fits into active site	
			3	ref. to (some enzymes) induced fit mechanism; A described	
			4	formation of enzyme-substrate complex; AW	
			5	lowering, activation energy / Ea; A detail of how activation energy lowered e.g substrates held close together for bond formati facilitates transfer of electrons places strain on bond(s) to be broken	[max 3]
		(ii)	1	loss of tertiary structure / hydrogen bonds broken / ionic bonds broken ; R if include disulfide or peptide bonds	
			2	changes shape / substrate unable to fit, active site; A enzyme changes salters active site	shape so
			3	loss of / AW, globular structure;	
			4	hydrophobic groups to outside of molecule;	
			5	hydrophilic groups no longer interact with water / AW;	[max 2]

- (e) penalise once for no units
 - with no cryoprotectant, enzyme (remains), inactive / AW;
 A at 0 mmol of cryoprotectant, 0% (of maximum) activity
 - 2 for both, increasing concentration increases % (enzyme) activity recovered;
 A comparative data quote with ref. to increase need units
 - 3 trehalose, steeper curve / AW, up to 10 mmol (cryoprotectant); ora R rapid
 - 4 at all concentrations (below 90 mmol), trehalose has higher percentage of (maximum enzyme) activity
 - 5 comparative data quote to support either mps 3 or 4; for mp 3 trehalose from 0 to 80% and glycerol from 0 to 10%
 - 6 both cryoprotectants can produce, 100% / maximum, (enzyme, activity / recovery);
 - 7 trehalose produces, 100% (enzyme) activity / full (enzyme) recovery at, lower concentrations than glycerol / 30 mmol compared to, 90-100 (mmol); this is also mp 6
 - trehalose more effective than glycerol (up to 90 95 mmol cryoprotectant);
 A trehalose is a better cryoprotectant (than glycerol)
 [max 4]

[Total: 16]

CHEMISTRY ONLINE

THITION —

4 (a put ticks and crosses against the boxes
1 – 4 and 7 – one letter only – if more than one letter mark as wrong
allow two or three correct letters for 5
allow two correct letters for 6

	statement	letter
1	contains peptide bonds	Н
2	part of the molecule forms the hydrophobic part of cell membranes	L
3	contains 1-4 and 1-6 glycosidic bonds	K
4	forms the primary structure of a protein	Н
5	used for energy storage in plants	K/M/H
6	forms a helical structure	M/H
7	the sub-unit molecule is β-glucose	J

[Total: 7]

