

Enzymes

Question Paper 7

Level	International A Level
Subject	Biology
Exam Board	CIE
Topic	Enzymes
Sub Topic	Enzymes
Booklet	Theory
Paper Type	Question Paper 7

Time Allowed : 64 minutes

Score : / 53

Percentage : /100

Grade Boundaries:

A*	A	B	C	D	E	U
>85%	77.5%	70%	62.5%	57.5%	45%	<45%

- 1 Starch is composed of two polysaccharides, amylose and amylopectin.

Fig. 3.1 shows a molecule of α -glucose before being added to the end of a molecule of amylose.

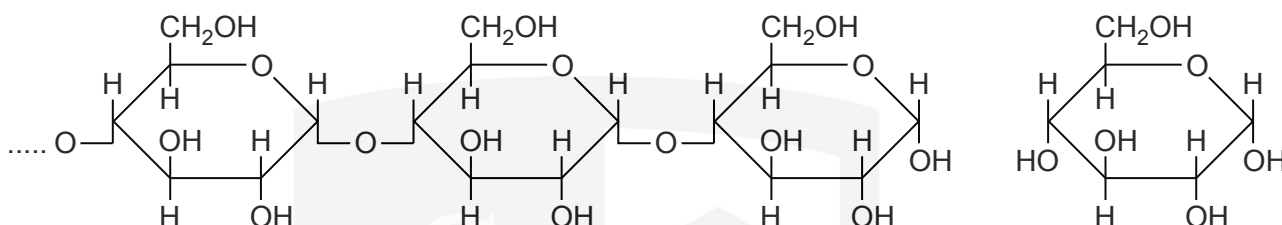


Fig. 3.1

- (a) (i) Complete Fig. 3.1 to show how a molecule of α -glucose is added to the amylose. [3]

- (ii) Name the bond that forms between glucose molecules in polysaccharides, such as amylose.

.....[1]

- (b) Glycogen and cellulose are two other polysaccharides.

Complete Table 3.1 to compare glycogen and cellulose with amylose.

Table 3.1

feature	amylose	glycogen	cellulose
monomer	α -glucose		
branched or unbranched molecule	unbranched		
role in organisms	energy storage		

(c) Type 2 diabetes (insulin-independent diabetes) is a non-infectious disease.

If not treated, this disease is characterised by large fluctuations in the concentration of glucose in the blood.

Maltase is an enzyme that completes the digestion of starch in humans. Molecules of maltase are bound to the microvilli of epithelial cells in the small intestine.

Ascorbase is a drug used in the treatment of type 2 diabetes. Molecules of ascorbase have a very similar shape to that of the substrate for maltase.

(i) Explain how ascorbase acts to inhibit these membrane-bound enzymes.

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..... [3]

(ii) Suggest why ascorbase can be used to treat people who have type 2 diabetes.

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..... [2]

[Total: 12]

2 Polysaccharides are synthesised by condensation reactions between monosaccharide or disaccharide subunits (monomers).

(a) Name the type of bond formed when polysaccharides are synthesised.

..... [1]

(b) Disaccharides are formed following synthesis from monosaccharides or as a result of polysaccharide hydrolysis.

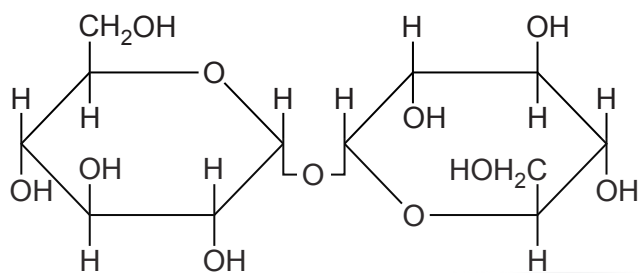
Cellobiose, maltose, sucrose and trehalose are four different disaccharides found in nature. Fig. 4.1 shows the molecular structure of these disaccharides.

Identify the disaccharides, labelled **A** to **D**, using the information below.

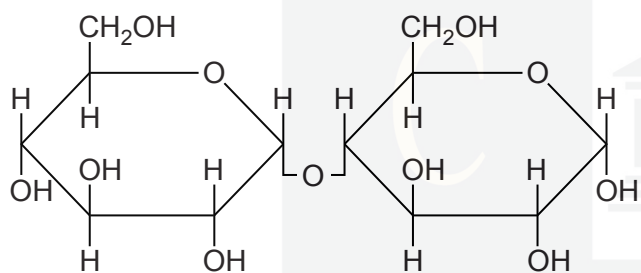
- The disaccharide cellobiose is formed from the hydrolysis of the polysaccharide cellulose.
- When cellobiose is hydrolysed, two **β -glucose** molecules are produced.
- One of the disaccharides is sucrose.
- Trehalose is a disaccharide that is synthesised from two **α -glucose** molecules.
- The disaccharide maltose is formed from the hydrolysis of amylose, a component of starch.

Write the name of the disaccharides in the spaces provided on Fig. 4.1.

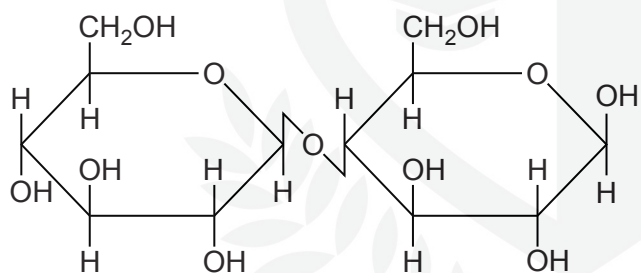
CHEMISTRY ONLINE
— TUITION —



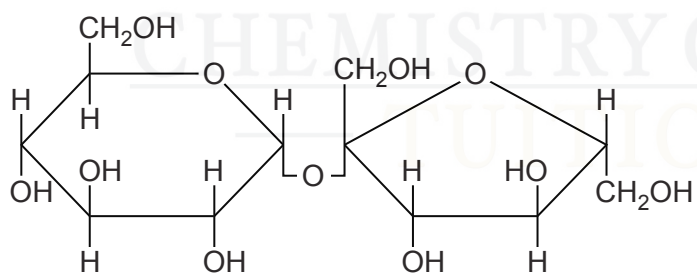
A



B



C



D

[3]

Fig. 4.1

In some organisms, trehalose is used as an energy store and gives protection against the harmful effects of very low temperatures. Trehalose is sometimes referred to as a cryoprotectant, allowing organisms to survive in freezing conditions.

Freezing temperatures can damage the cell surface membrane and membranes within the cell.

(c) Explain the importance of the cell surface membrane to cells.

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..... [3]

(d) Freezing temperatures can also completely stop enzyme activity by causing the molecules to undergo 'cold denaturation'. Enzyme activity is not recovered when temperatures are increased to a normal working temperature range.

(i) Explain the mode of action of enzymes.

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..... [3]

(ii) Suggest how the molecular structure of the enzyme changes during 'cold denaturation'.

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..... [2]

3 Resistance to the poison warfarin is now extremely common in rats. Warfarin inhibits an enzyme in the liver, vitamin K epoxide reductase, that is necessary for the recycling of vitamin K. This vitamin is involved in the production of substances required for blood clotting.

- Rats susceptible to warfarin die of internal bleeding.
- Rats that are homozygous for resistance to warfarin do not suffer from internal bleeding when their diet provides more than $70\mu\text{g}$ of vitamin K per kg body mass per day.
- Heterozygous rats are resistant to warfarin when their diet provides about $10\mu\text{g}$ of vitamin K per kg body mass per day.

(a) Using appropriate symbols, complete the genetic diagram to show how two resistant rats can produce warfarin-susceptible offspring.

key to symbols

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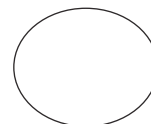
parental phenotypes

resistant male

resistant female

parental genotypes

gametes



offspring genotypes

offspring phenotypes

[3]

(b) Rats that are homozygous for warfarin resistance have a low survival rate in the wild. Suggest why this is so.

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..... [1]

- (c) Warfarin can be safely given to humans who are at risk of unwanted blood clots. The clotting time of the blood is measured regularly and the warfarin dose is varied accordingly.

Suggest, giving a reason, the type of inhibition warfarin has on the enzyme vitamin K epoxide reductase.

type of inhibition

reason

[2]

- (d) The allele for warfarin resistance may have originated by a single base substitution and resulted in a modified vitamin K epoxide reductase.

Explain how a single base substitution may affect the phenotype of an organism.

[3]

CHEMISTRY ONLINE
— TUITION —

[Total: 9]

- 4 (a) Carbon dioxide is transported in the blood in various forms.

Describe how carbon dioxide molecules reach red blood cells from respiring cells.

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..... [2]

Fig. 2.1 shows part of a capillary network and some cells of the surrounding tissue.

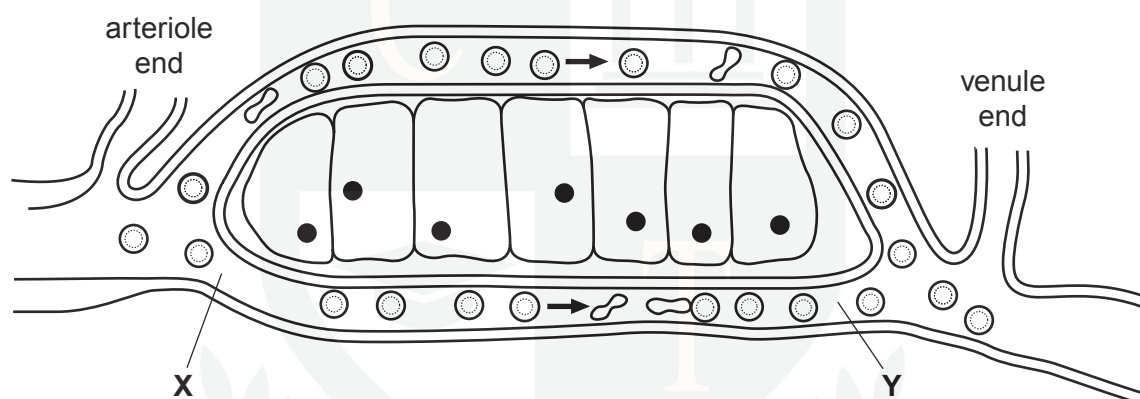


Fig. 2.1

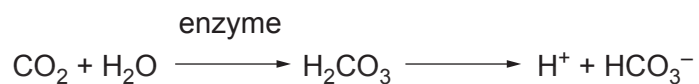
- (b) State three ways in which the blood at Y differs from the blood at X **other than** in the concentration of carbon dioxide.

1.

2.

3. [3]

An enzyme in red blood cells catalyses the reaction between carbon dioxide and water as blood flows through respiring tissues.



(c) (i) Name the enzyme that catalyses this reaction.

.....[1]

(ii) Explain the significance of this reaction in the transport of carbon dioxide.

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.....[3]

CHEMISTRY ONLINE
— TUITION —

- (d) Fig. 2.2 shows the effect of increasing the carbon dioxide concentration on the oxygen haemoglobin dissociation curve.

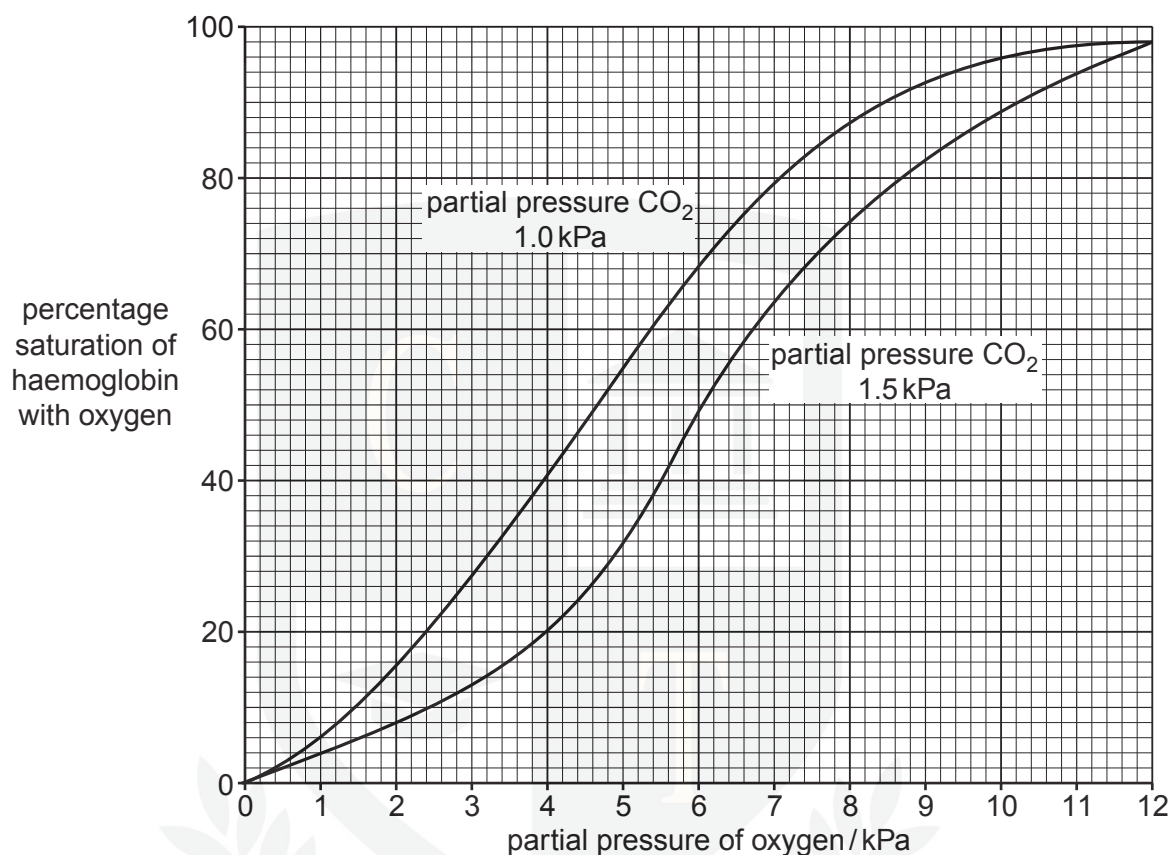


Fig. 2.2

- (i) State the percentage saturation of haemoglobin with oxygen at a partial pressure of 5 kPa of oxygen when the partial pressure of carbon dioxide is:

1.0 kPa

1.5 kPa [1]

- (ii) The percentage saturation of haemoglobin with oxygen **decreases** as the partial pressure of carbon dioxide increases.

Explain how this happens.

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..... [2]

- (iii) Name the effect of increasing carbon dioxide concentration on the oxygen dissociation curve.

.....[1]

- (iv) Explain the importance of the effect of carbon dioxide on haemoglobin as shown in Fig. 2.2.

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.....[3]

[Total: 16]

