

# Particle Physics

## Question paper 3

<b>Level</b>	International A Level
<b>Subject</b>	Physics
<b>Exam Board</b>	CIE
<b>Topic</b>	Particle & Nuclear Physics
<b>Sub Topic</b>	Particle Physics
<b>Paper Type</b>	Theory
<b>Booklet</b>	Question paper 3

**Time Allowed:** 75 minutes

**Score:** /62

**Percentage:** /100

CHEMISTRY ONLINE

A*	A	B	C	D	E	U
>85%	77.5%	70%	62.5%	57.5%	45%	<45%

- 1 Uranium-236 ( $^{236}_{92}\text{U}$ ) and Uranium-237 ( $^{237}_{92}\text{U}$ ) are both radioactive. Uranium-236 is an  $\alpha$ -emitter and Uranium-237 is a  $\beta$ -emitter.

(a) Distinguish between an  $\alpha$ -particle and a  $\beta$ -particle.

.....

.....

.....

.....

.....

..... [4]

- (b) The grid of Fig. 7.1 shows some proton numbers  $Z$  on the x-axis and the number  $N$  of neutrons in the nucleus on the y-axis.

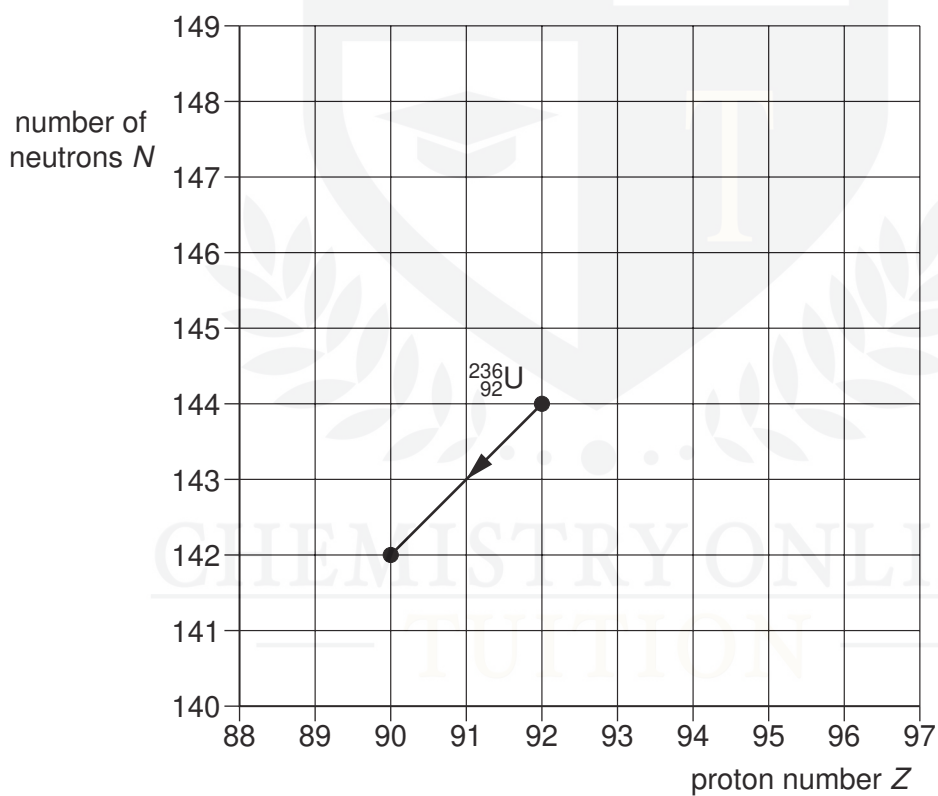


Fig. 7.1

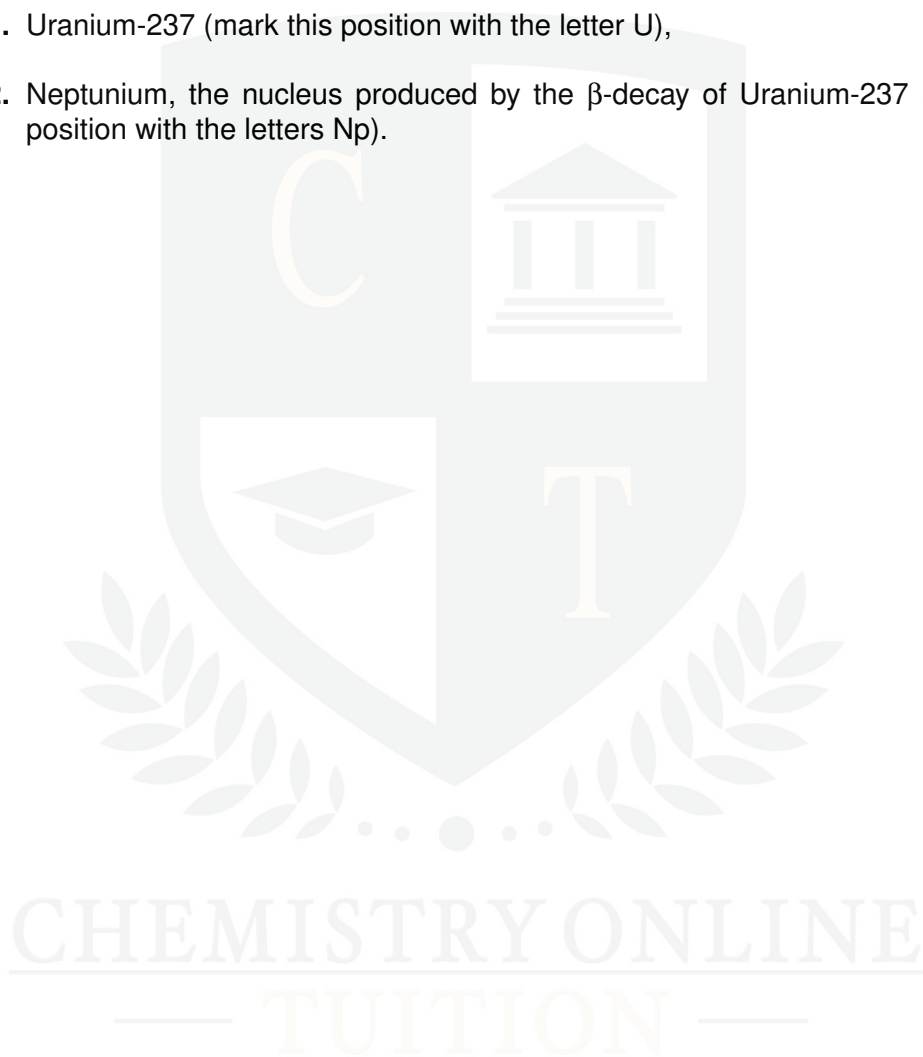
The  $\alpha$ -decay of Uranium-236 ( $^{236}_{92}\text{U}$ ) is represented on the grid. This decay produces a nucleus of thorium (Th).

(i) Write down the nuclear equation for this  $\alpha$ -decay.

..... [2]

(ii) On Fig. 7.1, mark the position for a nucleus of

1. Uranium-237 (mark this position with the letter U),
2. Neptunium, the nucleus produced by the  $\beta$ -decay of Uranium-237 (mark this position with the letters Np). [2]



- 2 (a) Evidence for the nuclear atom was provided by the  $\alpha$ -particle scattering experiment.  
State the results of this experiment.

.....

.....

.....

..... [2]

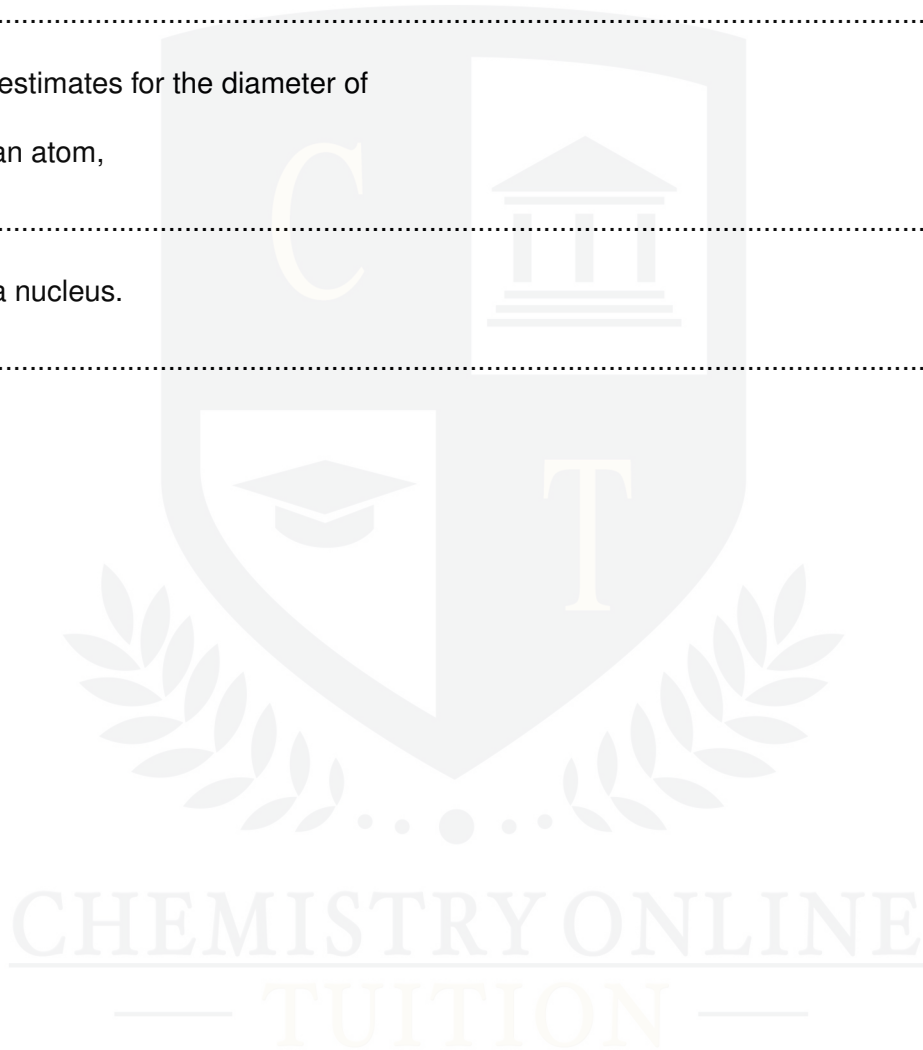
- (b) Give estimates for the diameter of

- (i) an atom,

.....[1]

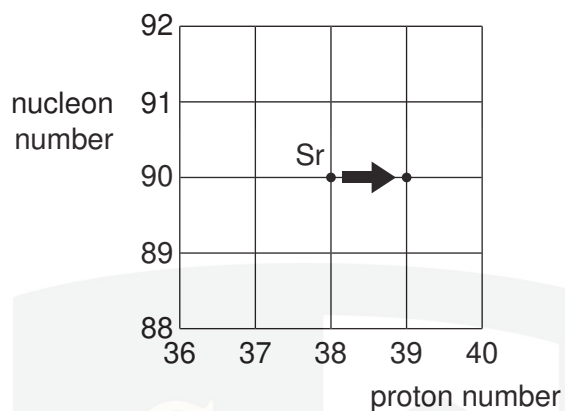
- (ii) a nucleus.

.....[1]





- 3 The radioactive decay of a strontium (Sr) nucleus is represented in Fig. 7.1.



**Fig. 7.1**

- (a) State whether Fig. 7.1 represents  $\alpha$ -decay,  $\beta$ -decay or  $\gamma$ -decay.

.....[1]

- (b) One type of radioactive decay cannot be represented on Fig. 7.1.  
Identify this decay and explain why it cannot be represented.

.....  
.....  
.....[2]

CHEMISTRY ONLINE  
— TUITION —

4 The  $\alpha$ - particle scattering experiment provided evidence for the existence of a nuclear atom.

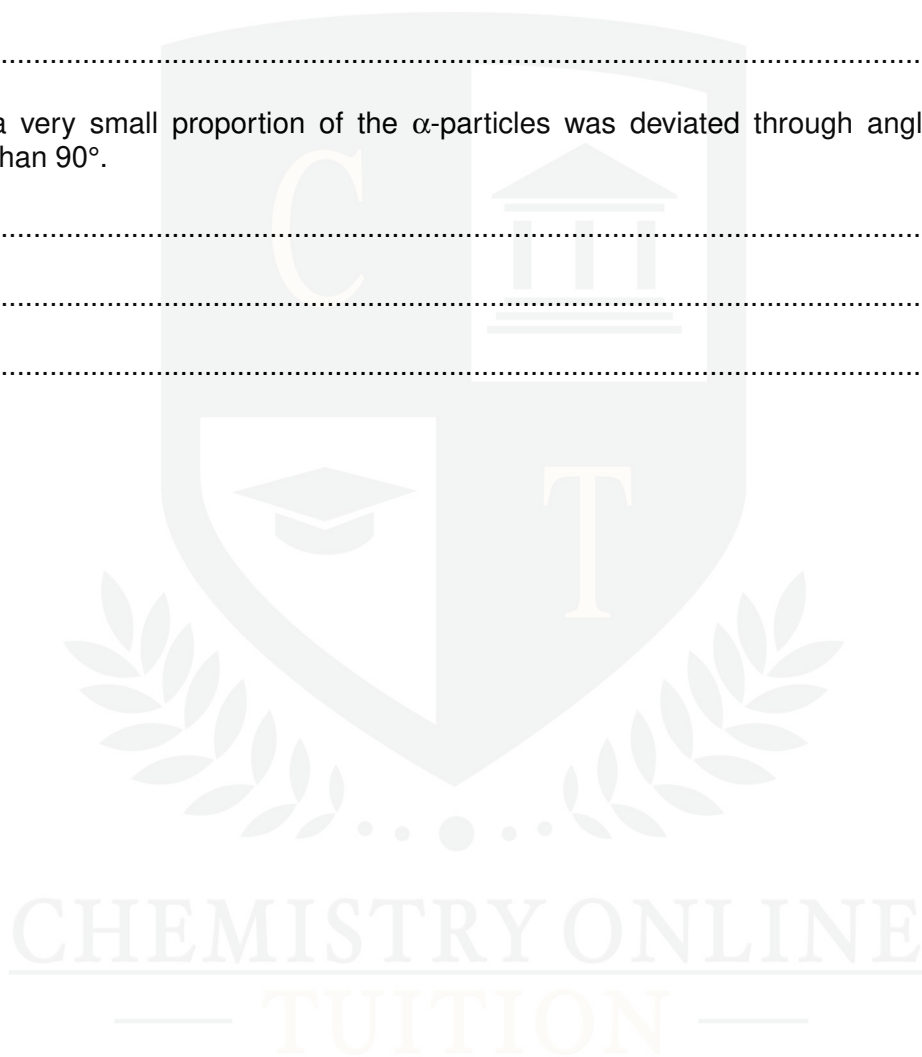
(a) State what could be deduced from the fact that

(i) most  $\alpha$ -particles were deviated through angles of less than  $10^\circ$ ,

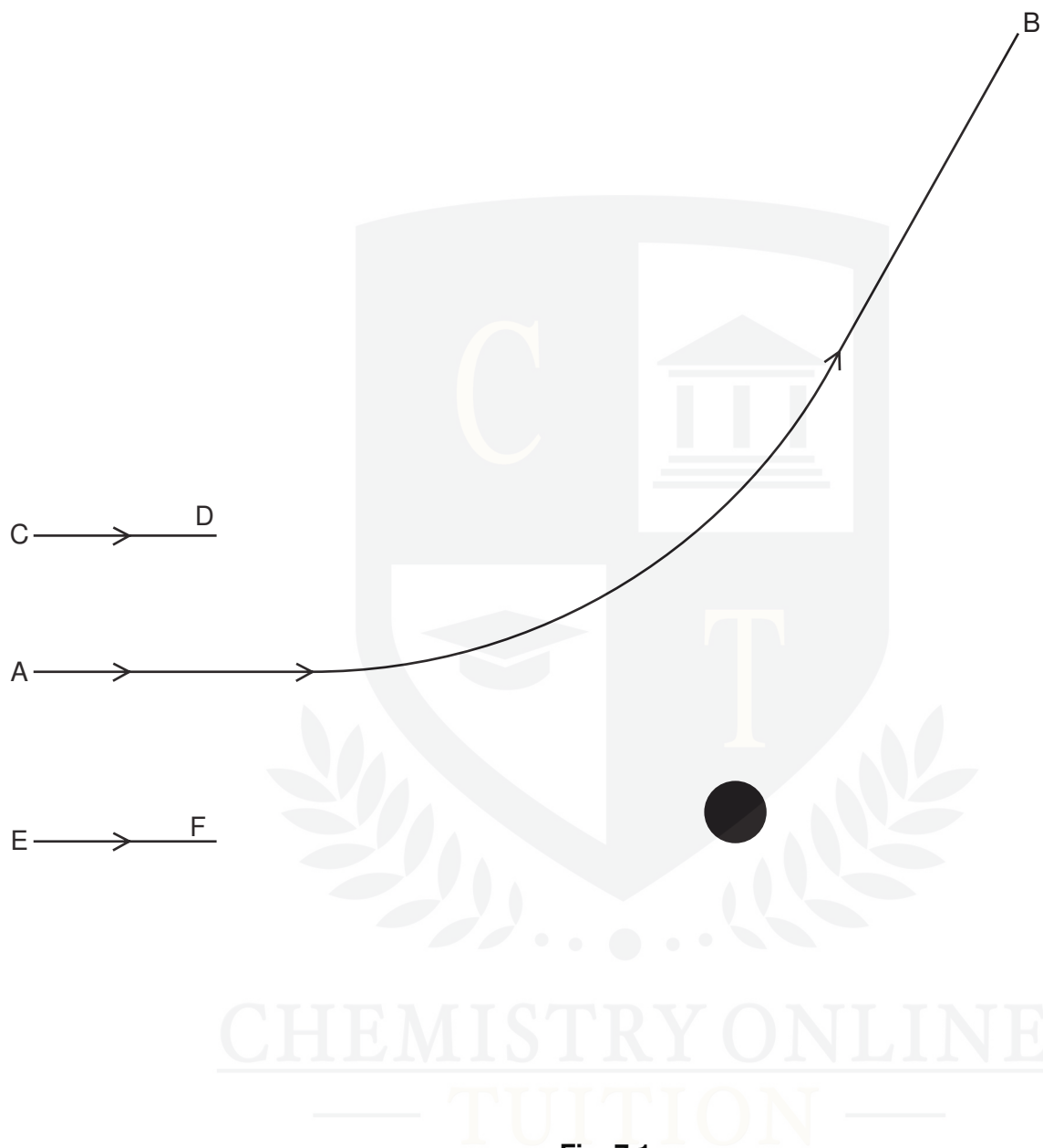
.....  
.....  
..... [2]

(ii) a very small proportion of the  $\alpha$ -particles was deviated through angles greater than  $90^\circ$ .

.....  
.....  
..... [2]



- (b) Fig. 7.1 shows the path AB of an  $\alpha$ -particle as it approaches and passes by a stationary gold nucleus.



**Fig. 7.1**

On Fig. 7.1, draw lines (one in each case) to complete the paths of the  $\alpha$ -particles passing by the gold nucleus when the initial direction of approach is

- (i) along line CD,
- (ii) along line EF.

[3]

5 One isotope of iron may be represented by the symbol



(a) State, for one nucleus of this isotope,

(i) the number of protons,

number = .....

(ii) the number of neutrons.

number = .....

[2]

(b) The nucleus of this isotope of iron may be assumed to be a sphere of radius  $5.7 \times 10^{-15} \text{ m}$ .

Calculate, for one such nucleus,

(i) the mass,

mass = ..... kg

(ii) the density.

density = .....  $\text{kg m}^{-3}$   
[4]

- (c) An iron ball is found to have a density of  $7900 \text{ kg m}^{-3}$ . By reference to your answer in (b)(ii), suggest what can be inferred about the structure of an atom of iron.

.....

.....

..... [2]



**6** A nucleus of an atom of francium (Fr) contains 87 protons and 133 neutrons.

**(a)** Write down the notation for this nuclide.

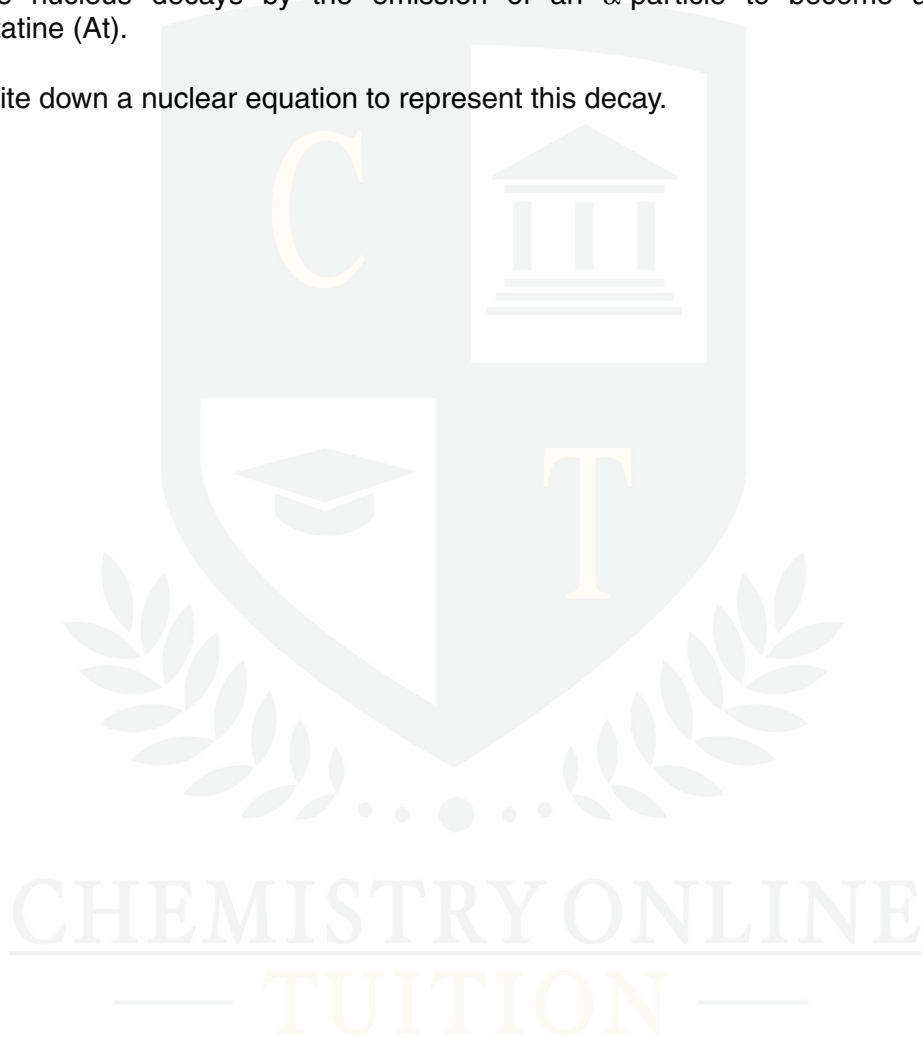
.....  
Fr  
.....

[2]

**(b)** The nucleus decays by the emission of an  $\alpha$ -particle to become a nucleus of astatine (At).

Write down a nuclear equation to represent this decay.

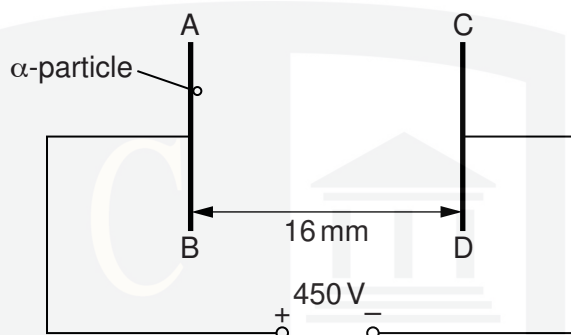
[2]



- 7 (a) Explain what is meant by an *electric field*.

.....  
.....[1]

- (b) A uniform electric field is produced between two vertical metal plates AB and CD, as shown in Fig. 7.1.



**Fig. 7.1**

The potential difference between the plates is 450V and the separation of the plates is 16 mm.

An  $\alpha$ -particle is accelerated from plate AB to plate CD.

- (i) On Fig. 7.1, draw lines to represent the electric field between the plates. [2]  
(ii) Calculate the electric field strength between the plates.

electric field strength = .....  $\text{V m}^{-1}$  [2]

- (iii) Calculate the work done by the electric field on the  $\alpha$ -particle as it moves from AB to CD.

work done = ..... J [3]

(iv) A  $\beta$ -particle moves from AB to CD. Calculate the ratio

$$\frac{\text{work done by the electric field on the } \alpha\text{-particle}}{\text{work done by the electric field on the } \beta\text{-particle}}.$$

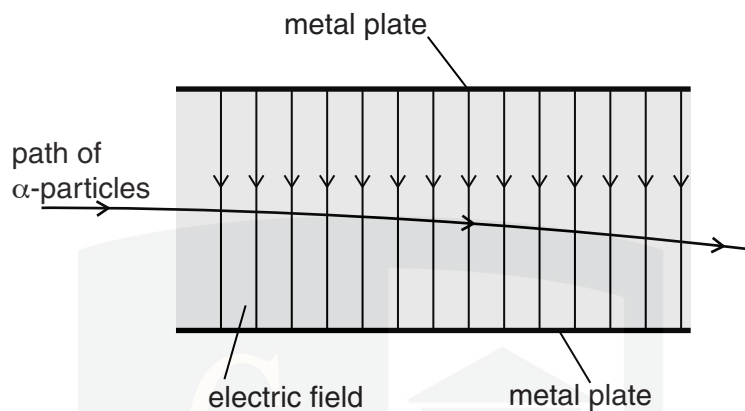
Show your working.

ratio = ..... [1]



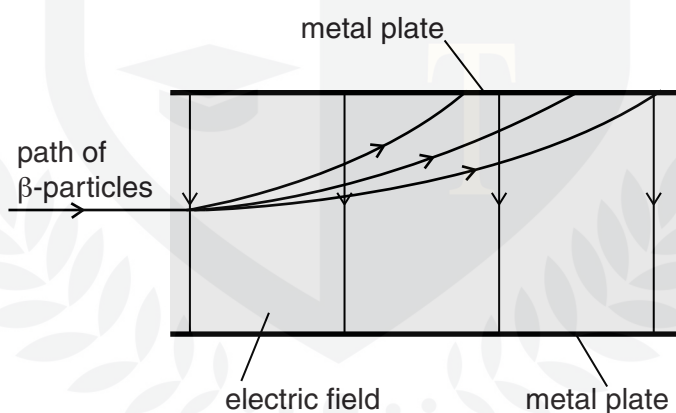


- 8 (a) An electric field is set up between two parallel metal plates in a vacuum. The deflection of  $\alpha$ -particles as they pass between the plates is shown in Fig. 7.1.



**Fig. 7.1**

The electric field strength between the plates is reduced. The  $\alpha$ -particles are replaced by  $\beta$ -particles. The deflection of  $\beta$ -particles is shown in Fig. 7.2.



**Fig. 7.2**

- (i) State one similarity of the electric fields shown in Fig. 7.1 and Fig. 7.2.

.....  
.....[1]

- (ii) The electric field strength in Fig. 7.2 is less than that in Fig. 7.1. State two methods of reducing this electric field strength.

1. ....  
2. ....

[2]

- (iii) By reference to the properties of  $\alpha$ -particles and  $\beta$ -particles, suggest three reasons for the differences in the deflections shown in Fig. 7.1 and Fig. 7.2.

1. ....  
.....
2. ....  
.....
3. ....  
.....
- [3]

- (b) A source of  $\alpha$ -particles is uranium-238. The nuclear reaction for the emission of  $\alpha$ -particles is represented by



State the values of  $W$ .....

$X$  .....

$Y$  .....

$Z$  .....

[2]

- (c) A source of  $\beta$ -particles is phosphorus-32. The nuclear reaction for the emission of  $\beta$ -particles is represented by



State the values of  $A$  .....

$B$  .....

$C$  .....

$D$  .....

[1]

- 9 Two horizontal metal plates are separated by distance  $d$  in a vacuum. A potential difference  $V$  is applied across the plates, as shown in Fig. 6.1.

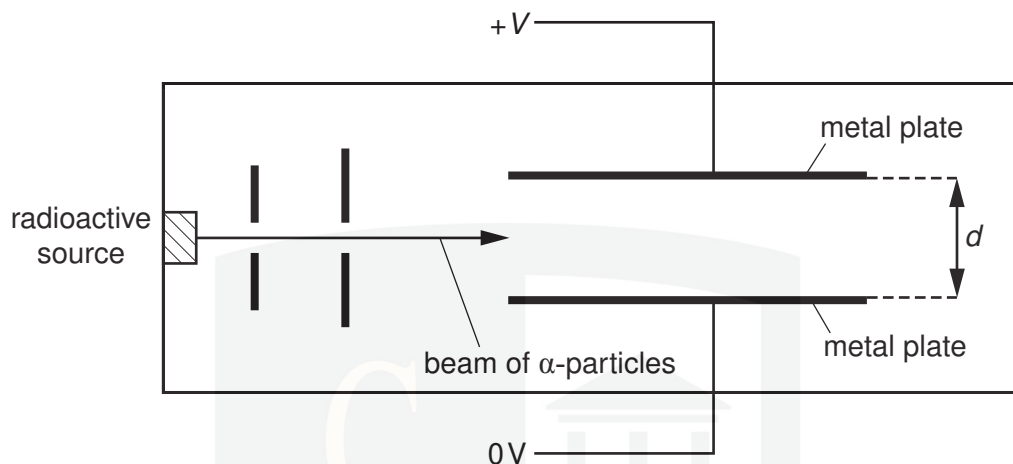


Fig. 6.1

A horizontal beam of  $\alpha$ -particles from a radioactive source is made to pass between the plates.

- (a) State and explain the effect on the deflection of the  $\alpha$ -particles for each of the following changes:

- (i) The magnitude of  $V$  is increased.

.....  
..... [1]

- (ii) The separation  $d$  of the plates is decreased.

.....  
..... [1]

- (b) The source of  $\alpha$ -particles is replaced with a source of  $\beta$ -particles. Compare, with a reason in each case, the effect of each of the following properties on the deflections of  $\alpha$ - and  $\beta$ -particles in a uniform electric field:

(i) charge

.....  
.....  
.....[2]

(ii) mass

.....  
.....  
.....[2]

(iii) speed

.....  
.....  
.....[1]

- (c) The electric field gives rise to an acceleration of the  $\alpha$ -particles and the  $\beta$ -particles. Determine the ratio

$$\frac{\text{acceleration of the } \alpha\text{-particles}}{\text{acceleration of the } \beta\text{-particles}}$$

ratio = .....[3]