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## **CHEMISTRY**

## PHYSICAL CHEMISTRY

LEVEL & BOARD:	OCR (AS - LEVEL)
TOPIC:	Compounds, Formulae & Equations
PAPER TYPE:	SOLUTION 4
TOTAL QUESTIONS	07
TOTAL MARKS	17

## Compounds, Formula, Equation -4

a.l

• Iodide ion

Q.2

• Equation between Radium and Oxygen

$$2 Ra + O_2 \rightarrow 2 RaO$$

Q.3

 Molecular formula of alkane with 22 carbons Molecular formula

C22H46

• So, Empirical formula would be

Q.4

• Reaction of Aluminum with chlorine

$$2AI + CI_2 \rightarrow 2AICI_3$$

Q.5

• Silver chloride in the presence of sunlight decomposer to give metallic silver which is grey in colour,

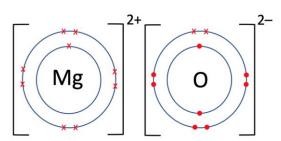
$$\begin{array}{ccc} & \text{Sunlight} \\ 2 \text{Agcl}_{(s)} & \longrightarrow & 2 \text{Ag}_{(s)} + \text{Cl2}_{(g)} \end{array}$$

(b)

Magnesium hydroxide is used to treat acidity and constipation

(ii) 
$$Mg(s)_1 + SO_{2(g)} \rightarrow MgO(s) + S_8(s)$$

• (iii) Dot and cross diagram between magnesium oxide



**'Q.6** 

• Ions present in Mg3(PO4)2 are

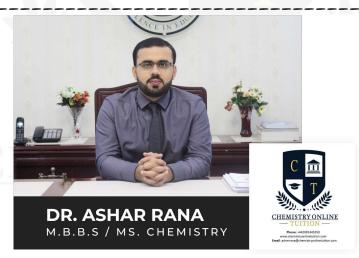
$$PO_{\Lambda}^{2-}$$
 Anion

**Q.7** 

• Strontium reaction with water

$$Sr_s + H_2O_{(e)} \rightarrow Sr(OH)_2_{eq} + H_2(g)$$

- (ii) Sr<sub>3</sub>N<sub>2</sub> has ionic bonding present in it.
  - (iii) Dot and cross diagram for SrCl2



- · Founder & CEO of Chemistry Online Tuition Ltd.
- · Completed Medicine (M.B.B.S) in 2007
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